

# Electronic Structure and Mechanical Properties of 20 MAX-Phase Compounds

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# LARGE SCALE SIMULATIONS OF THE MECHANICAL PROPERTIES OF LAYERED TRANSITION METAL TERNARY COMPOUNDS FOR FOSSIL ENERGY POWER SYSTEM APPLICATIONS

*NETL Project DE-FE0005865 (Technical Manager: Richard Dunst)*

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**Co-PIs: Paul Rulis and Lizhi Ouyang**

*25<sup>th</sup> Annual Conference on Fossil Energy Materials*

*Portland, Oregon, April 26-28, 2011*

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- I. Background of the project**
- II. Project outline and significance**
- III. Statement of project objectives (OSPO)**
- IV. The project team**
- V. Technical approach and methods used**
- VI. Some preliminary results**

# What I present today at the 26<sup>th</sup> ACFEM is essentially a Report

## OUTLINE:

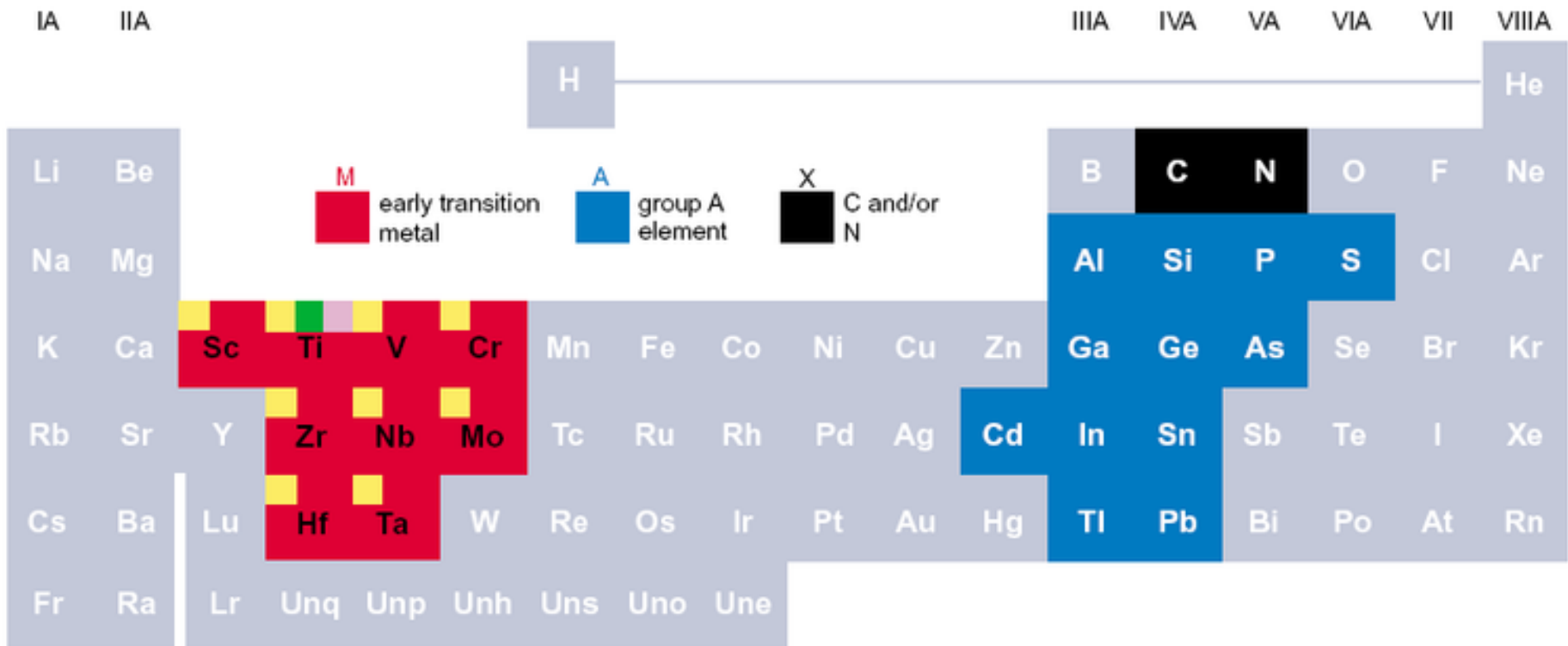
1. Introduction to MAX phase compounds
2. Why MAX compounds
- 3. Results on 20 MAX phases**
  - a. Electronic and optical properties**
  - b. Mechanical and elastic properties**
4. Implication of results
5. Summary
6. Work in progress or planned

# Introduction to MAX Phase

**What is MAX?** Layered Ternary Transition Metal Carbides and Nitrides

**Formula:**  $M_{n+1}AX_n$  where **M** — Early Transition Metal; **A** — A-group element; **X** — Carbide or Nitride,  $n = 2-5$ .  $n=1$  also possible.

Most phases are 211 or 312 compounds; 413 and 514 are very rare.

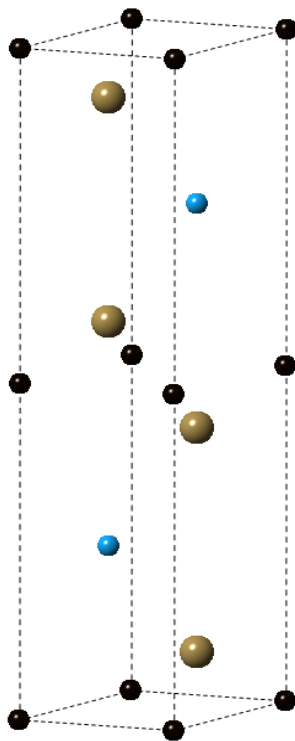


List of 70 MAX Phases			
(1 1 1) Phases	(2 1 1) Phases	(3 1 2) Phases	"Irregular" Phases
ScAlC	Ti <sub>2</sub> AlC	Ti <sub>3</sub> SiC <sub>2</sub>	HfAl <sub>3</sub> C <sub>3</sub>
VAlC	Ti <sub>2</sub> AlN	Ti <sub>3</sub> SiC <sub>2</sub>	Hf <sub>2</sub> Al <sub>3</sub> C <sub>4</sub>
TiAlC	Ti <sub>2</sub> SiC	Ti <sub>3</sub> AlC <sub>2</sub>	Hf <sub>3</sub> Al <sub>3</sub> C <sub>5</sub>
CrAlC	Ti <sub>2</sub> GeC	Ti <sub>3</sub> GeC <sub>2</sub>	Hf <sub>2</sub> Al <sub>4</sub> C <sub>5</sub>
ZrAlC	Ti <sub>2</sub> SnC	Ta <sub>3</sub> AlC <sub>2</sub>	Hf <sub>3</sub> Al <sub>4</sub> C <sub>6</sub>
NbAlC	Ti <sub>2</sub> PbC	Ti <sub>3</sub> SnC <sub>2</sub>	ZrAl <sub>3</sub> C <sub>3</sub>
MoAlC	Ti <sub>2</sub> PC	α-V <sub>3</sub> SiC <sub>2</sub>	Zr <sub>2</sub> Al <sub>3</sub> C <sub>4</sub>
HfAlC	Ti <sub>2</sub> SC	β-V <sub>3</sub> SiC <sub>2</sub>	Zr <sub>3</sub> Al <sub>3</sub> C <sub>5</sub>
WAlC	Ti <sub>2</sub> GaC	(4 1 3) Phases	Zr <sub>2</sub> Al <sub>4</sub> C <sub>5</sub>
TaAlC	Ti <sub>2</sub> InC	α-Ta <sub>4</sub> AlC <sub>3</sub>	Zr <sub>3</sub> Al <sub>4</sub> C <sub>6</sub>
	Ti <sub>2</sub> TiC	β-Ta <sub>4</sub> AlC <sub>3</sub>	Zr <sub>3</sub> La <sub>3</sub> C <sub>5</sub>
	Ti <sub>2</sub> AsC	Ti <sub>4</sub> AlN <sub>3</sub>	ZrAl <sub>4</sub> C <sub>4</sub>
	Cr <sub>2</sub> AlC	V <sub>4</sub> AlC <sub>3</sub>	Zr <sub>2</sub> Al <sub>3</sub> C <sub>5</sub>
	Cr <sub>2</sub> SiC	Nb <sub>4</sub> AlC <sub>3</sub>	γ-Y <sub>2</sub> Si <sub>2</sub> O <sub>7</sub>
	Cr <sub>2</sub> PC	(5 1 4) Phases	Al <sub>4</sub> SiC <sub>4</sub>
	Cr <sub>2</sub> SC	Ta <sub>5</sub> AlC <sub>4</sub>	La <sub>2</sub> Zr <sub>2</sub> O <sub>7</sub>
	V <sub>2</sub> AlC	(6 1 5) Phases	Cr <sub>0.5</sub> Al <sub>0.5</sub> N
	V <sub>2</sub> SiC	Ta <sub>6</sub> AlC <sub>5</sub>	Solid Solutions
	V <sub>2</sub> PC		Zr <sub>2</sub> [Al(Si)] <sub>4</sub> C <sub>5</sub>
	V <sub>2</sub> SC		Zr <sub>3</sub> [Al(Si)] <sub>4</sub> C <sub>6</sub>
	Nb <sub>2</sub> AlC		Ti <sub>3</sub> Si <sub>0.75</sub> Al <sub>0.25</sub> C <sub>2</sub>
	Nb <sub>2</sub> AsC		Ti <sub>3</sub> Si <sub>0.9</sub> Al <sub>0.1</sub> C <sub>2</sub>
	Ta <sub>2</sub> AlC		Ti <sub>3</sub> Si(Al)C <sub>2</sub>

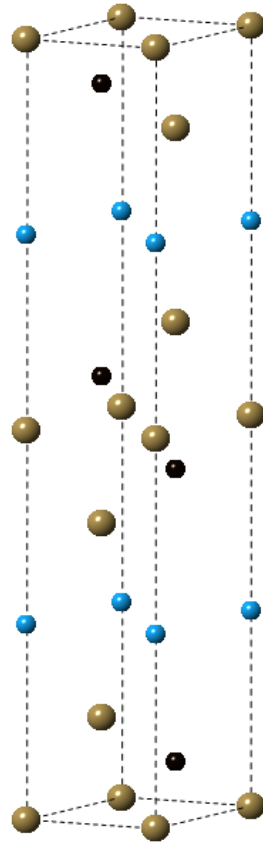
Many phases have not been synthesized. This list is rapidly increasing!

# Crystal structures of 211, 312, 413, 514 MAX phases

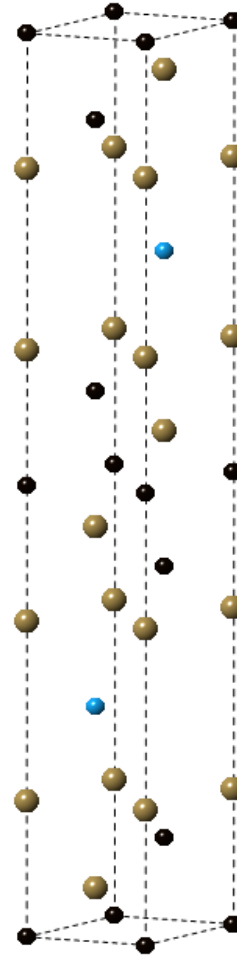
Ta Al C



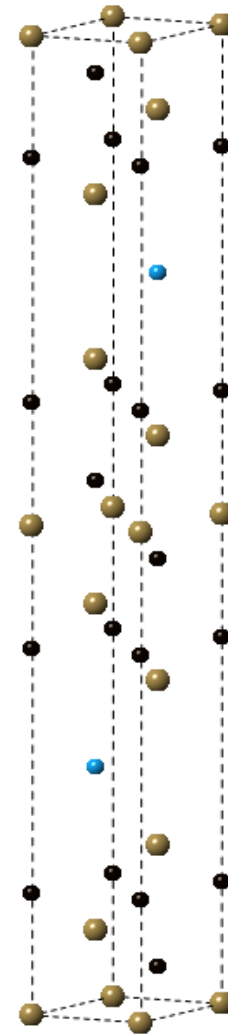
Ta<sub>2</sub>AlC



$\alpha$ -Ta<sub>3</sub>AlC<sub>2</sub>



$\alpha$ -Ta<sub>4</sub>AlC<sub>3</sub>

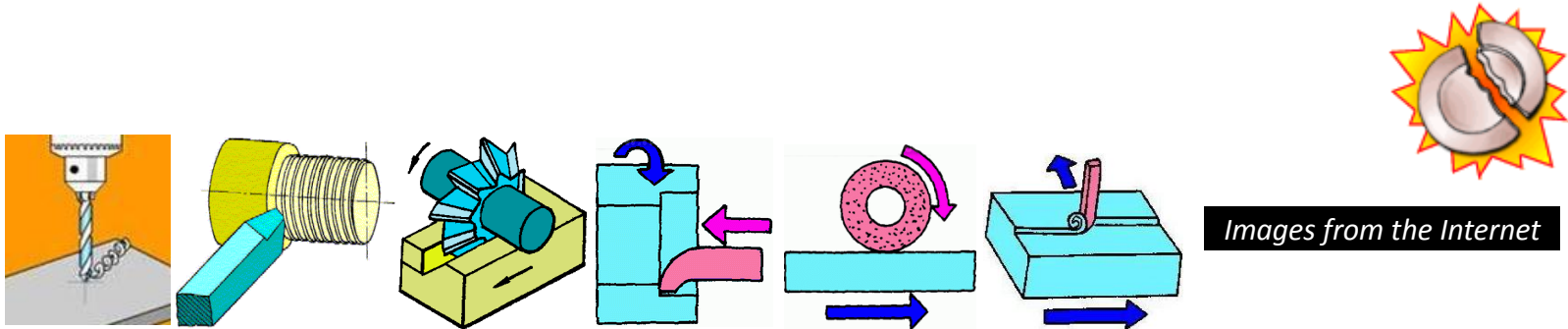


Ta<sub>5</sub>AlC<sub>4</sub>

# Special properties of MAX phases compounds

## Advanced Properties:

***Like a Metal:*** thermally and electrically conductive, thermal-shock resistant, machinable, and damage tolerant;



***Like Ceramics:*** light weighted, stiff, refractory and oxidation resistant, not too expensive.

**Mechanical properties of MAX phases are very complicated!!**

**Some are very promising, others are detrimental, depending on structure, composition and interatomic bonding.**

## Applications (realized and projected):

- High-temperature structural applications;
- Porous exhaust gas filters for automobiles;
- Heat exchangers;
- Heating elements;
- Wear and corrosion protective surface coatings;
- Electrodes, resistors, capacitors, rotating electrical contacts;
- Nuclear applications;
- Bio-compatible materials;
- As sensor materials;
- Cutting tools, nozzles, tools for die pressing;
- Impact-resistant materials; projectile proof armor, bullet proof vest, etc.

***As Advanced Materials in Fossil energy power plants?***

***This is the one of the main goals for this project!***

***==> Using ab initio computation to explore new materials systems for advance applications based on fundamental understanding!***



# Strategy

- Select 20 MAX phases of different components and compositions. They are:  $Ti_3MC_2$  (M=Al; Si, Ge),  $Ti_2MC$  (M=Al, Ga, In; Si, Ge, Sn; P, As; S),  $Ti_2AlN$ ,  $M_2AlC$  (M=V, Nb, Cr) and  $Ta_{n+1}AlC_n$  (n=1~4). (Materials Genome approach!)
- Calculate the Electronic Structure, bonding and optical conductivities of these 20 MAX phases.
- Calculate the elastic and mechanical properties of the same 20 MAX phases.
- Find the property **trends** among them and predict properties of other MAX phases.
- Explore the **correlations** between properties, compositions and structures and search for new MAX phases.

# Methods (brief!)

## Electronic structure calculations:

First-principles orthogonalized linear combination of atomic orbitals (**OLCAO**) method. Density Functional Theory (DFT) based using LDA, very efficient for complex systems.

**Effective Charge  $Q^*$**  on each atom, **Bond Order** values for each pair of atoms, are evaluated according to:

$$Q_{\alpha}^* = \sum_i \sum_{n,occupied} \sum_{j,\beta} C_{i\alpha}^{m*} C_{j\beta}^m \langle b_{i\alpha}(\vec{k}, \vec{r}) | b_{j\beta}(\vec{k}, \vec{r}) \rangle$$

$$\rho_{\alpha\beta} = \sum_{n,occupied} \sum_{j,\beta} C_{i\alpha}^{n*} C_{j\beta}^n \langle b_{i\alpha}(\vec{k}, \vec{r}) | b_{j\beta}(\vec{k}, \vec{r}) \rangle$$

The real part of the **Optical Conductivity** is obtained by  $\sigma_1 = \frac{\varepsilon_2 \omega}{4\pi}$ , after evaluating

$$\varepsilon_2(\omega) = \left( \frac{e^2}{\pi m E \omega} \right) \times \int d\vec{k} \sum_{n,l} |\langle \psi_n(\vec{k}, \vec{r}) | \vec{P} | \psi_l(\vec{k}, \vec{r}) \rangle|^2 f_l(\vec{k}) \times [1 - f_n(\vec{k})] \delta[E_n(\vec{k}) - E_l(\vec{k}) - E].$$

## Mechanical properties Calculations:

Use Vienna *Ab initio* Simulation Package (**VASP**) with high accuracy, a stress-strain analysis of under linear elastic theory, and RVH approximation for poly-crystals.

**These methods have been successfully used by us in many crystals and non-crystalline materials in recent years.**

### 3. Results on 20 MAX phases

Show what have been accomplished so far, not into details for each of the 20 MAX phases compounds.

#### a. Electronic and optical properties

Band structures

Density of states (DOS) and Partial DOS (PDOS)

Effective charge and bond order values

Optical conductivities

XANES spectra

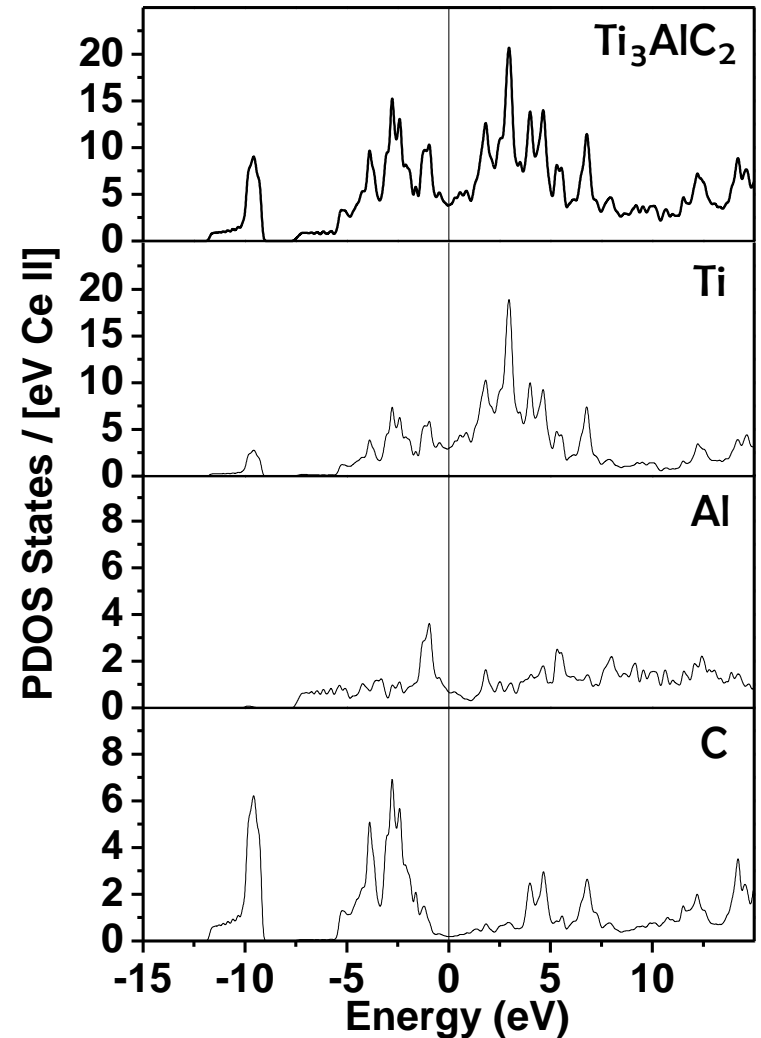
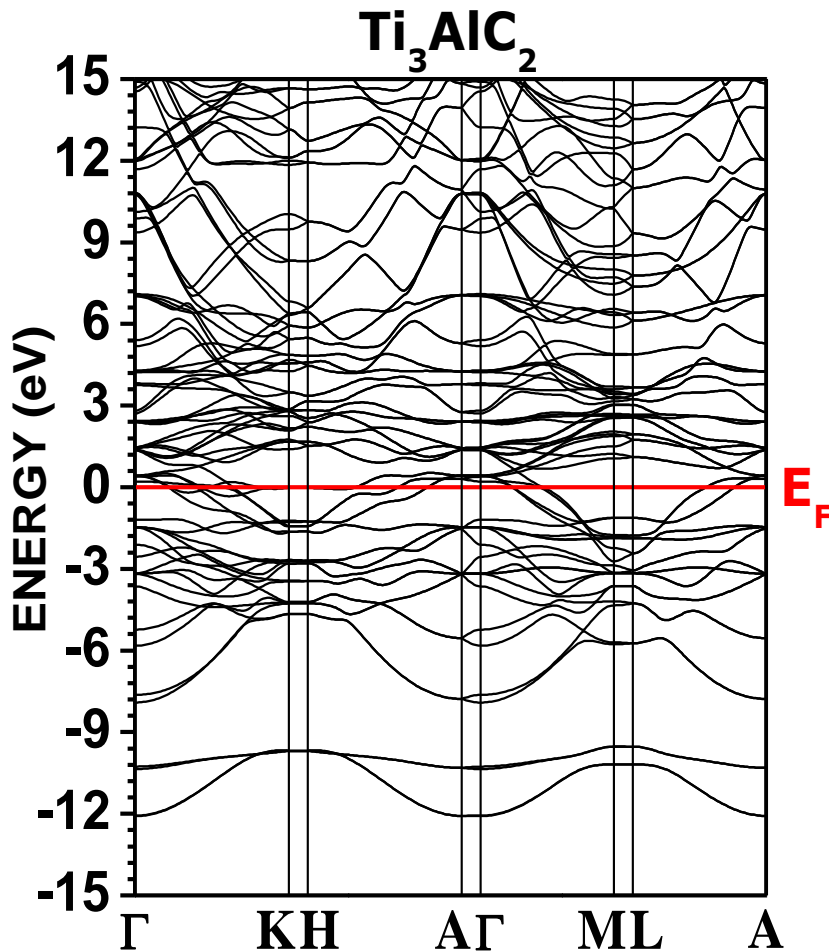
#### b. Mechanical and elastic properties.

Elastic coefficients

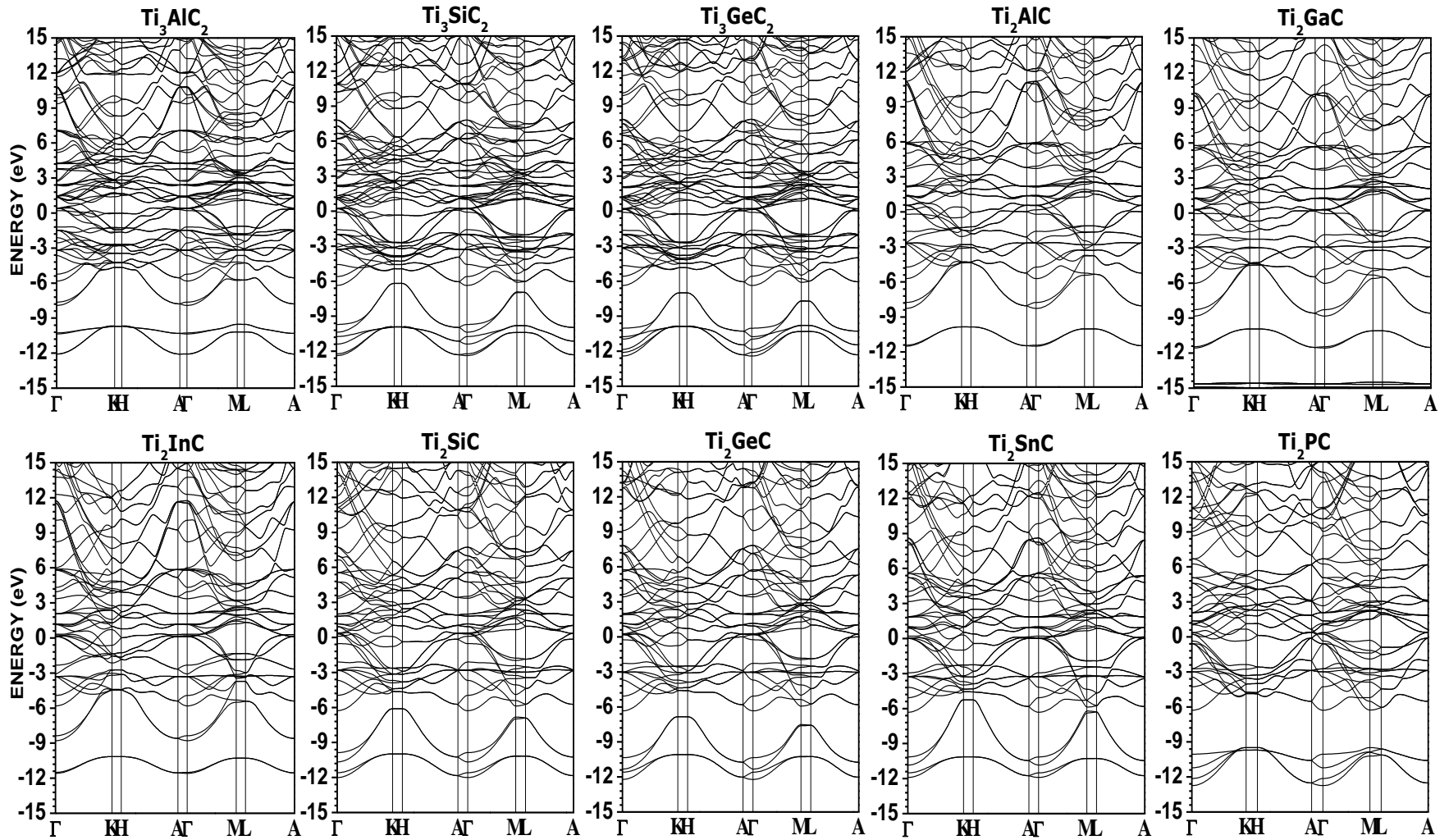
Mechanical parameters (K,G,E, $\eta$ ,  $k=G/K$ )

Additional analysis

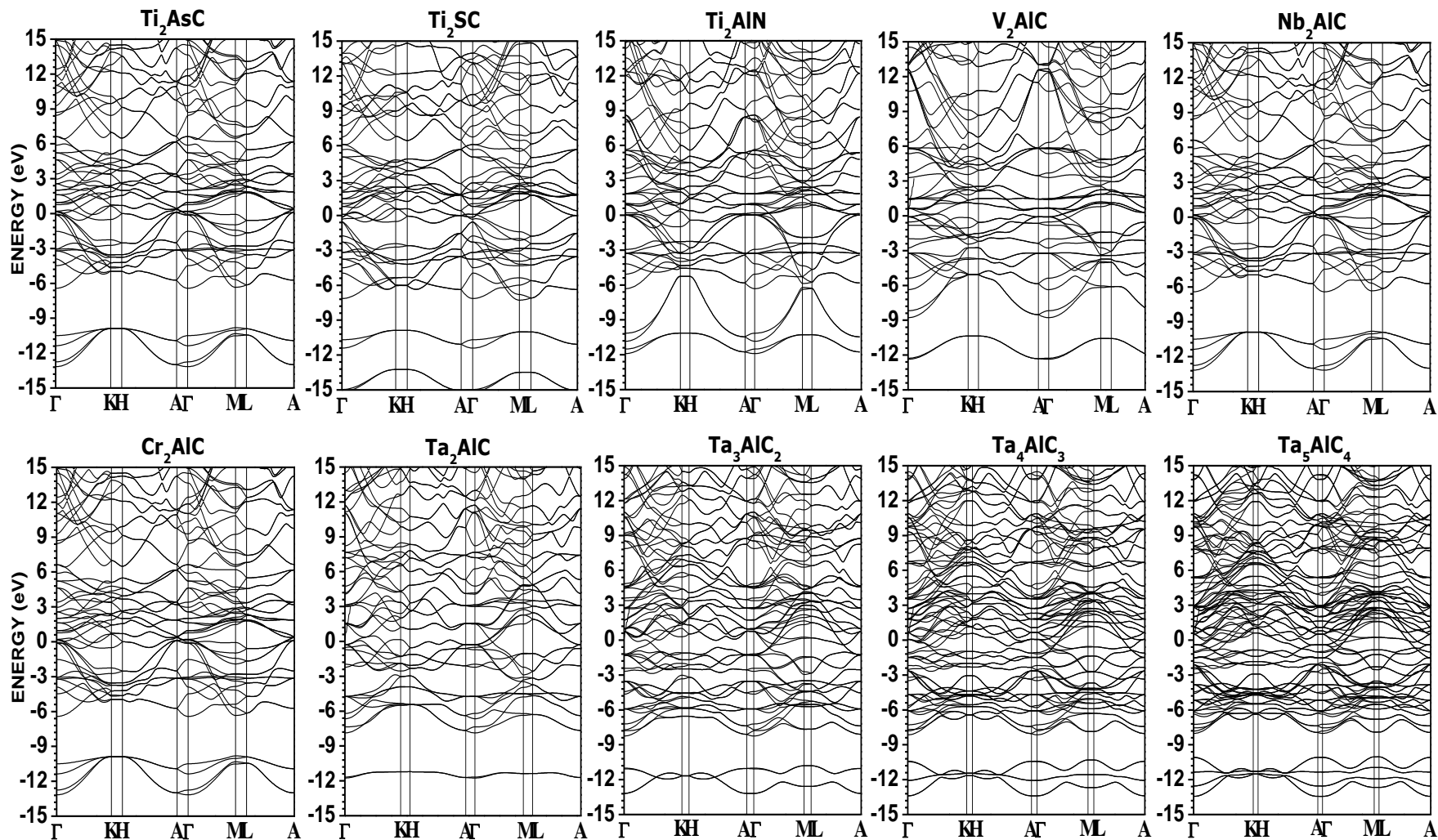
# Example on the results for the electronic structure of MAX: Band Structure and Density of states in $\text{Ti}_3\text{AlC}_2$



# Band structure of MAX phases: 1-10

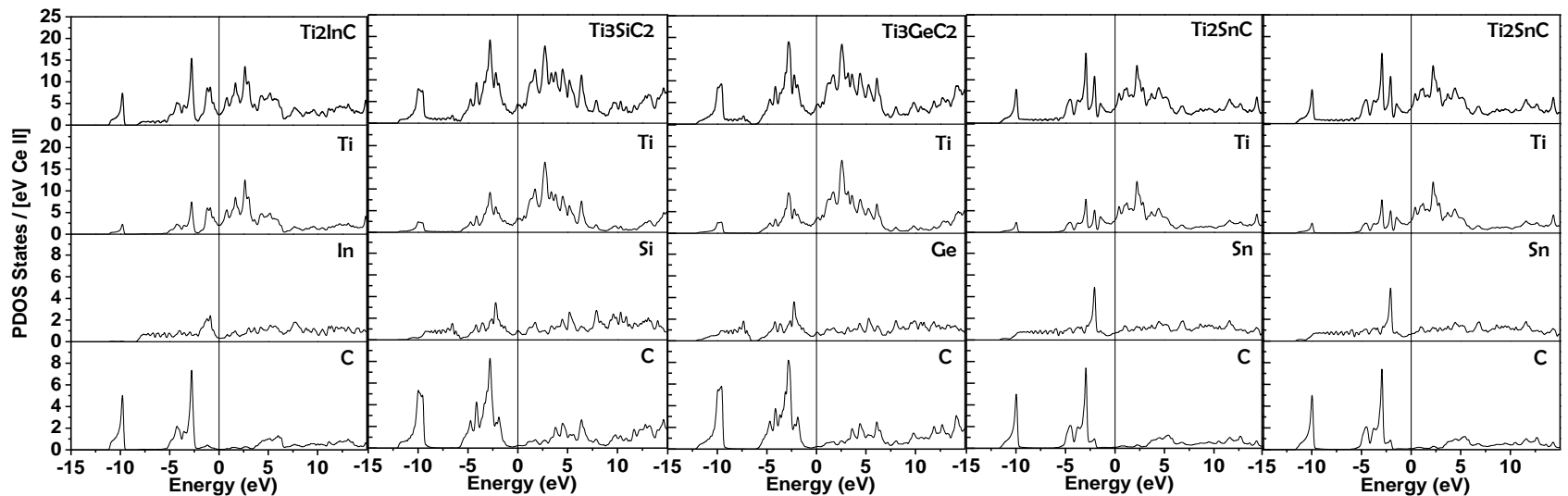
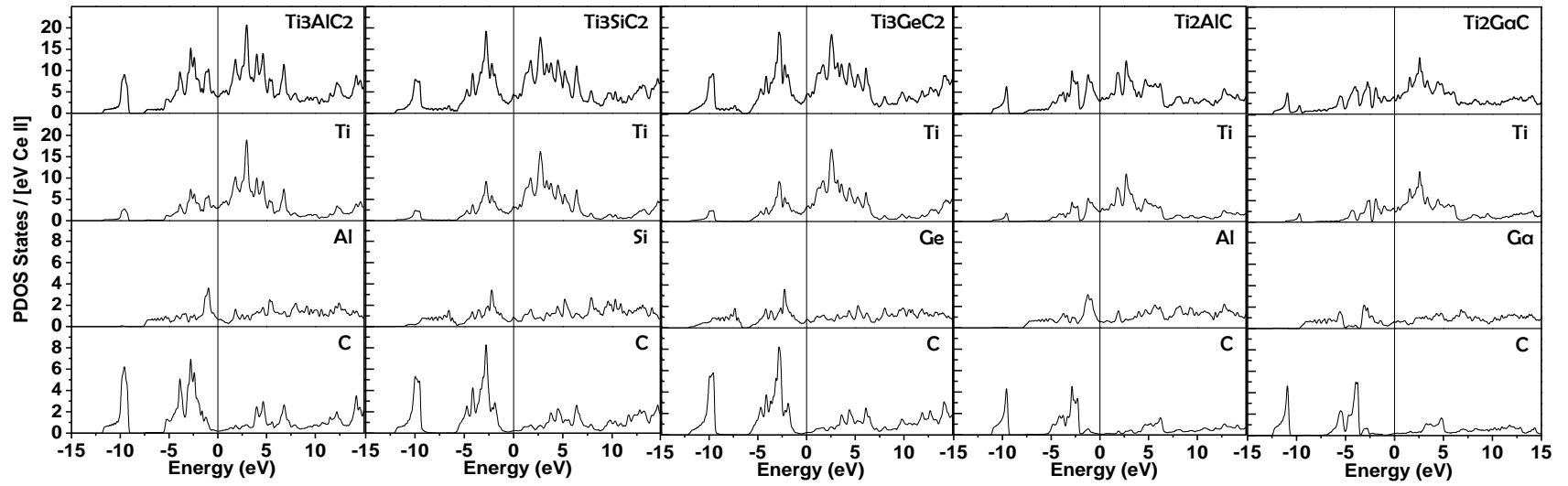


# Band structure of MAX phases: 11-20

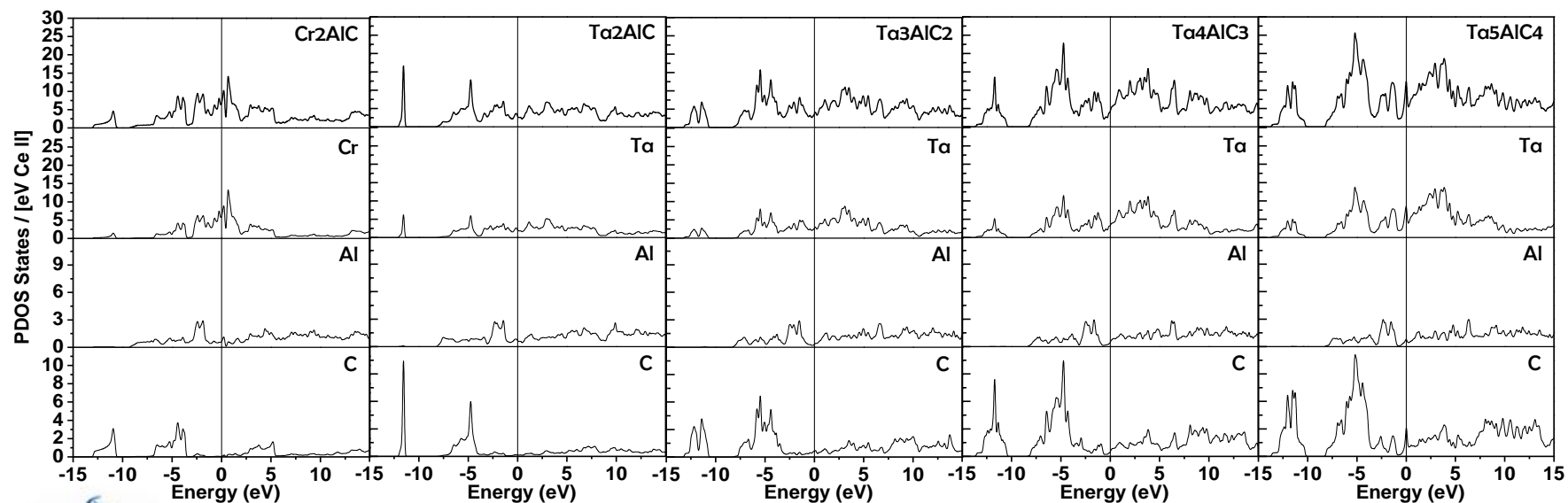
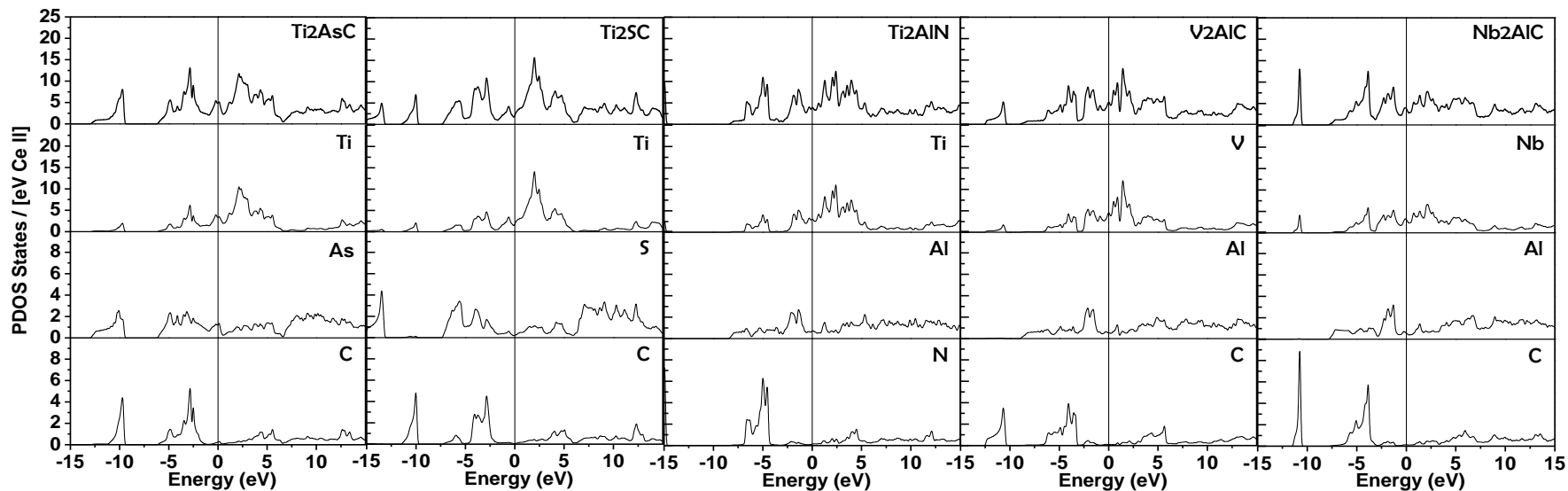




# DOS and PDOS of MAX phases: 1-10



# DOS and PDOS of MAX phases: 11-20

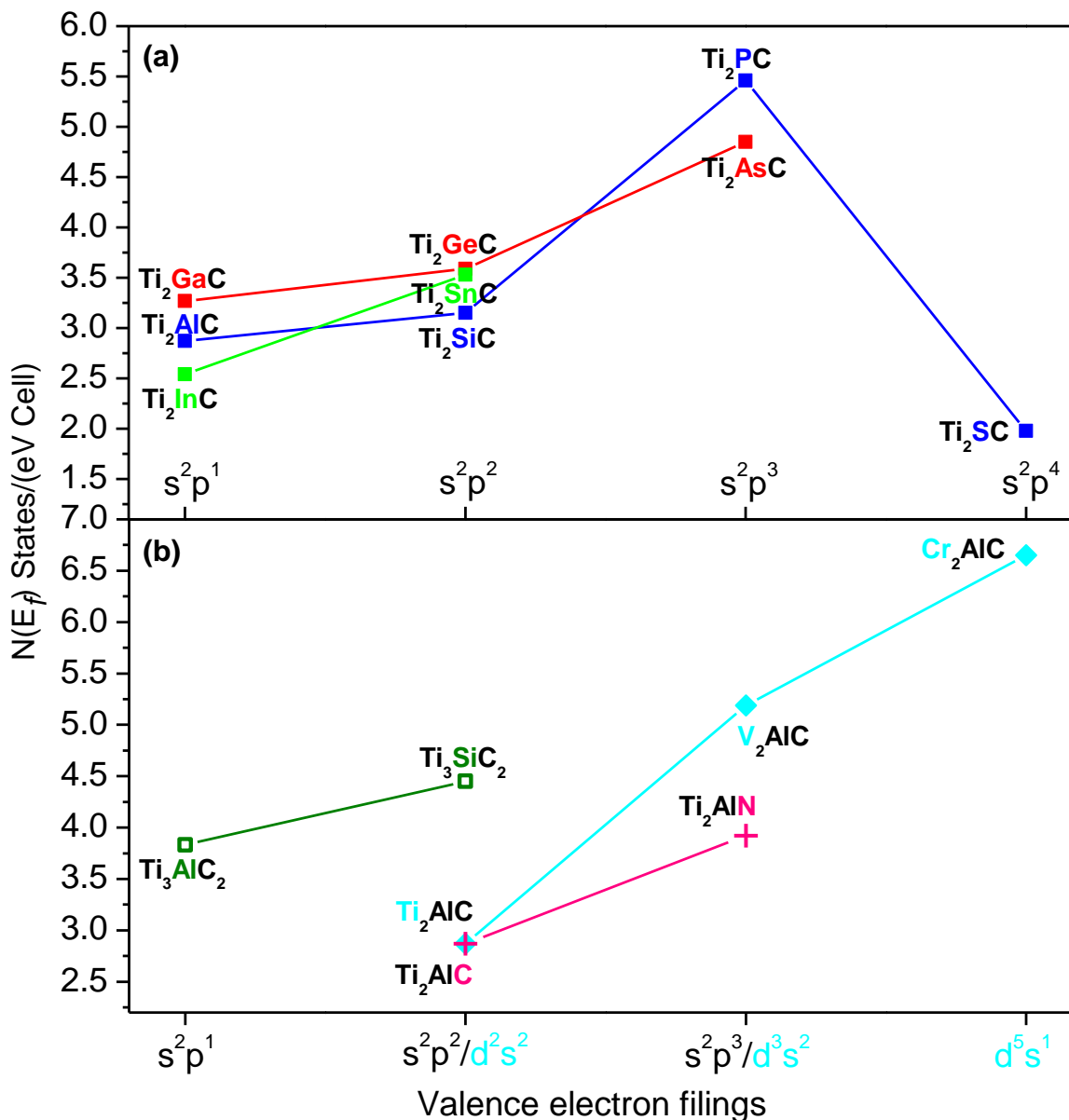




# Total and partial (atom-resolved) DOS at Fermi-level

Crystal	TDOS (Fermi level)	$N(E_f)$ M	$N(E_f)$ A	$N(E_f)$ X
Ti <sub>3</sub> AlC <sub>2</sub>	3.826	2.97	0.67	0.19
Ti <sub>3</sub> SiC <sub>2</sub>	4.453	3.32	0.89	0.24
Ti <sub>3</sub> GeC <sub>2</sub>	4.331	3.33	0.78	0.22
Ti <sub>2</sub> AlC	2.867	2.24	0.55	0.08
Ti <sub>2</sub> GaC	3.274	2.76	0.44	0.08
Ti <sub>2</sub> InC	2.537	2.14	0.34	0.06
Ti <sub>2</sub> SiC	3.152	2.36	0.69	0.10
Ti <sub>2</sub> GeC	3.592	2.76	0.73	0.10
Ti <sub>2</sub> SnC	3.527	2.80	0.63	0.10
Ti <sub>2</sub> PC	5.464	3.95	1.22	0.29
Ti <sub>2</sub> AsC	4.845	3.36	1.24	0.25
Ti <sub>2</sub> SC	1.981	1.72	0.20	0.06
Ti <sub>2</sub> AlN	3.915	3.06	0.72	0.14
V <sub>2</sub> AlC	5.191	4.56	0.53	0.10
Nb <sub>2</sub> AlC	3.838	3.14	0.57	0.13
Cr <sub>2</sub> AlC	6.647	5.96	0.63	0.06
Ta <sub>2</sub> AlC	2.918	2.08	0.68	0.16
Ta <sub>3</sub> AlC <sub>2</sub>	3.646	2.70	0.30	0.64
Ta <sub>4</sub> AlC <sub>3</sub>	4.217	3.24	0.48	0.50
Ta <sub>5</sub> AlC <sub>4</sub>	12.403	8.52	0.80	3.08

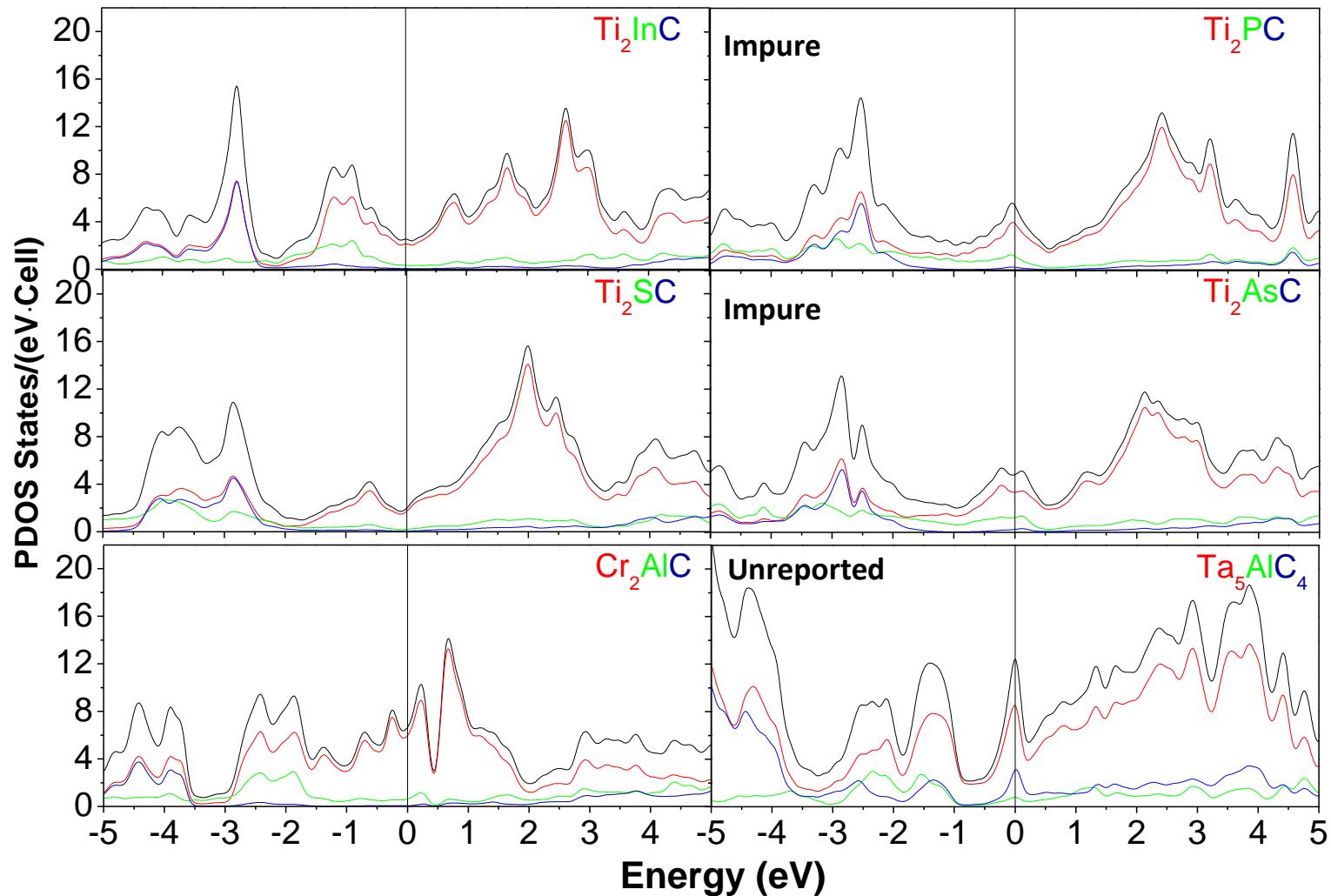
# The trend of $N(E_f)$ with respect to valence electron filings



# Prediction of Phase Stability: Peak or valley of DOS at $E_F$ .

Stable

Unstable

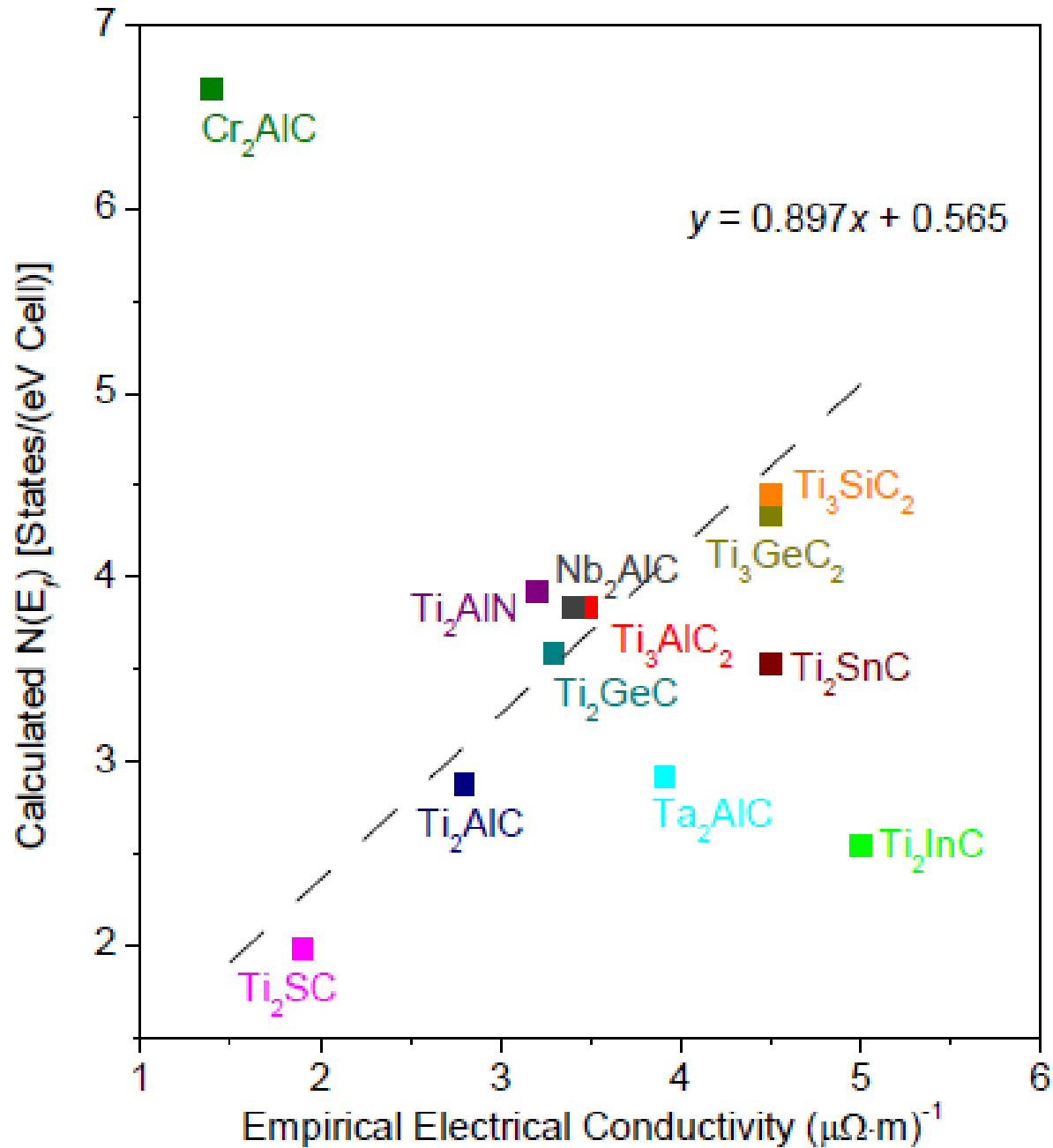


*Consistent with experimental observations!!*

# Correlation of $N(E_f)$ and Experimental Electrical Conductivities

Crystal	$N(E_f)$ [States/(eV·Cell)]	Electrical Conductivity ( $\mu\Omega^{-1}\cdot\text{m}^{-1}$ )	
$\text{Ti}_3\text{AlC}_2$	3.826	3.48	Acta Materialia 2002;50:3141
$\text{Ti}_3\text{SiC}_2$	4.453	4.5	Scripta Materialia 2006;54:841
$\text{Ti}_3\text{GeC}_2$	4.331	4.5	Scripta Materialia 1997;36:535
$\text{Ti}_2\text{AlC}$	2.867	2.8	Scripta Materialia 1997;36:535
$\text{Ti}_2\text{InC}$	2.537	5.0	J. Alloys Compd. 2002;340:173
$\text{Ti}_2\text{GeC}$	3.592	3.3	Solid State Communications 2008;146:498
$\text{Ti}_2\text{SnC}$	3.527	4.5	J. Euro. Ceramic Society 2000;20:2619
$\text{Ti}_2\text{SC}$	1.981	1.9	J. App. Phys. 2008;104:033502
$\text{Ti}_2\text{AlN}$	3.915	3.2	Scripta Materialia 1997;36:535
$\text{Nb}_2\text{AlC}$	3.838	3.4	Met. and Mater. Trans. A 2002;33A:2775
$\text{Cr}_2\text{AlC}$	6.647	1.4	Scripta Materialia 2006;54:841
$\text{Ta}_2\text{AlC}$	2.918	3.91	J. Euro. Ceramic Society 2008;28:1679

# The correlation between $N(E_f)$ and the electrical conductivity

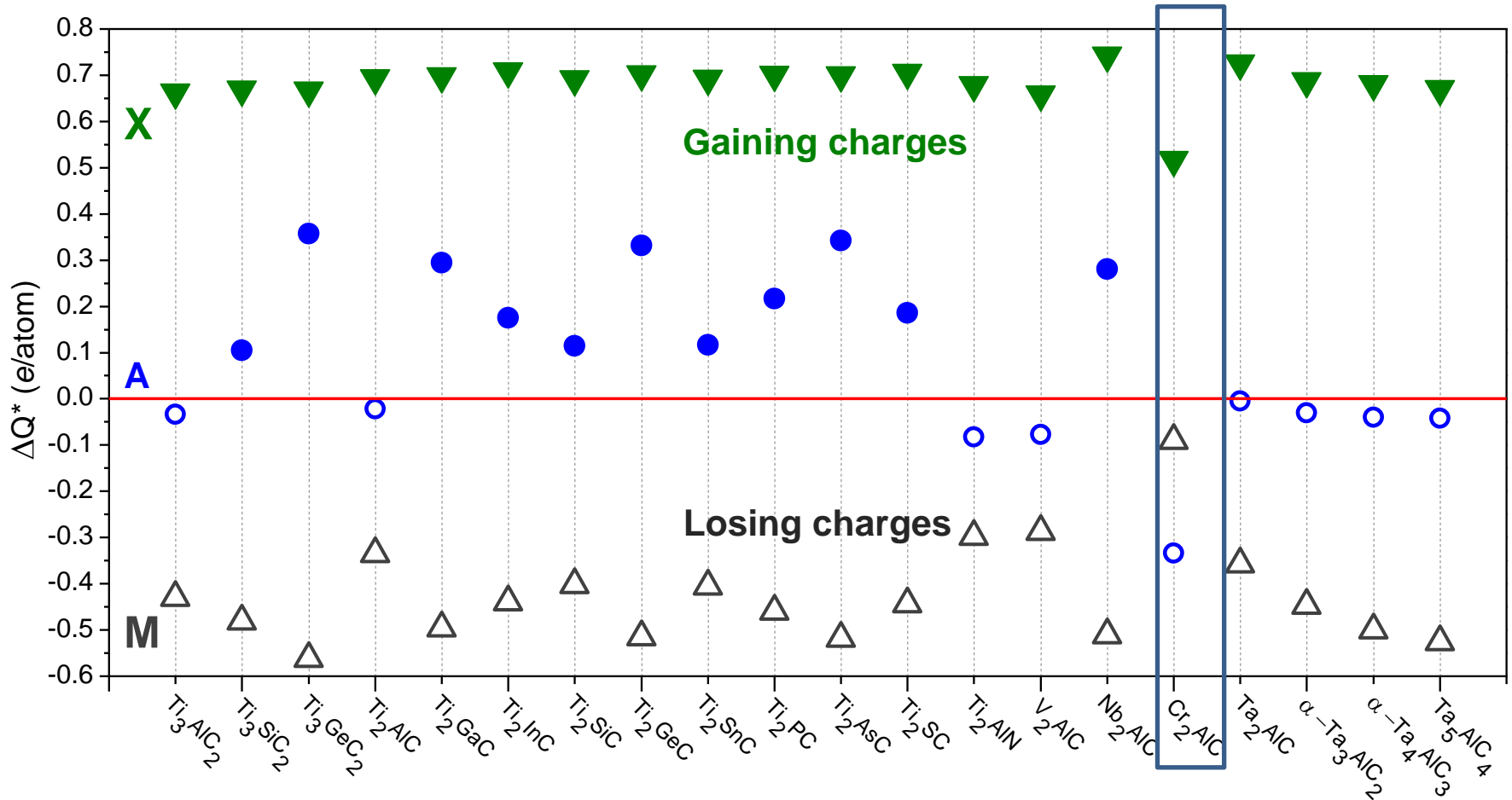


# Charge transfer in MAX phases from effective charge calculations

Crystal	$\Delta Q^*$ (e/atom)	$\Delta Q^*$ (e/atom)	$\Delta Q^*$ (e/atom)
	M	A	X
Ti <sub>3</sub> AlC <sub>2</sub>	-0.430	-0.034	0.663
Ti <sub>3</sub> SiC <sub>2</sub>	-0.481	0.105	0.669
Ti <sub>3</sub> GeC <sub>2</sub>	-0.562	0.357	0.667
Ti <sub>2</sub> AlC	-0.335	-0.022	0.694
Ti <sub>2</sub> GaC	-0.496	0.294	0.698
Ti <sub>2</sub> InC	-0.439	0.175	0.708
Ti <sub>2</sub> SiC	-0.402	0.114	0.691
Ti <sub>2</sub> GeC	-0.515	0.332	0.702
Ti <sub>2</sub> SnC	-0.405	0.117	0.693
Ti <sub>2</sub> PC	-0.459	0.217	0.701
Ti <sub>2</sub> AsC	-0.518	0.342	0.700
Ti <sub>2</sub> SC	-0.443	0.186	0.705
Ti <sub>2</sub> AlN	-0.298	-0.083	0.679
V <sub>2</sub> AlC	-0.287	-0.077	0.658
Nb <sub>2</sub> AlC	-0.511	0.280	0.742
Cr <sub>2</sub> AlC	-0.090	-0.334	0.517
Ta <sub>2</sub> AlC	-0.357	-0.005	0.725
$\alpha$ -Ta <sub>3</sub> AlC <sub>2</sub>	-0.447	-0.030	0.687
$\alpha$ -Ta <sub>4</sub> AlC <sub>3</sub>	-0.500	-0.040	0.681
Ta <sub>5</sub> AlC <sub>4</sub>	-0.525	-0.042	0.670

# The charge transfer between different composing elements:

M: lose charge; X: gain charge; A: mostly losing charge, some gain charge.



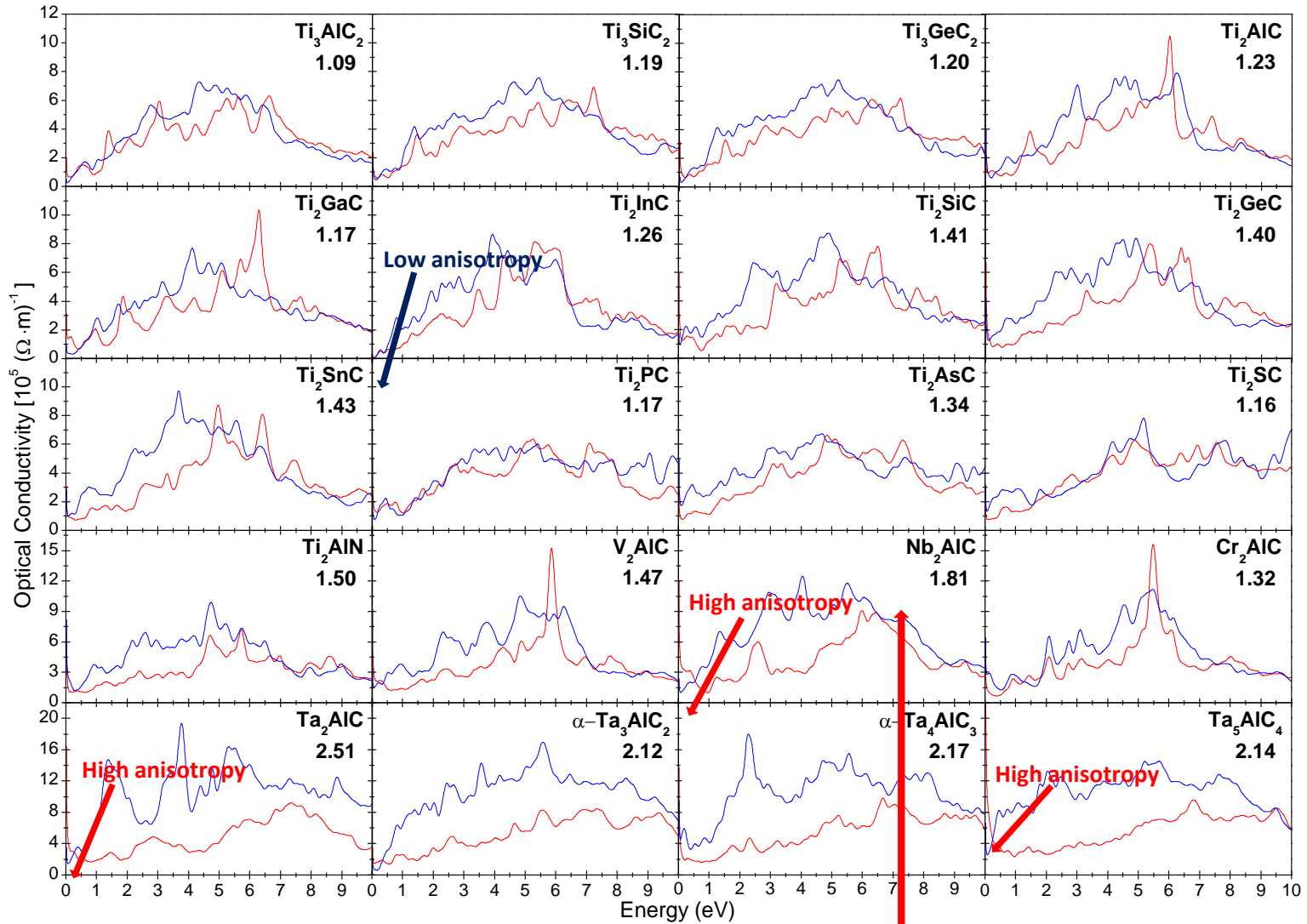
# Summary of bond order values for all the bonding types:

Very complicated! They are the fundamental controlling factor of their electronic and mechanical properties (*no time to discuss in detail here!*)

Crystal	M-X				M-A	M-M					A-A
						Intra-layer		Inter-layer			
								Cross X	Cross A		
Ti <sub>2</sub> AlC	0.212				0.159	0.069		0.050			0.063
Ti <sub>2</sub> GaC	0.213				0.148	0.058		0.055			0.058
Ti <sub>2</sub> InC	0.212				0.139	0.059		0.065			0.070
Ti <sub>2</sub> SiC	0.214				0.173	0.052		0.042			0.031
Ti <sub>2</sub> GeC	0.215				0.151	0.050		0.048			0.032
Ti <sub>2</sub> SnC	0.214				0.152	0.054		0.056			0.040
Ti <sub>2</sub> PC	0.216				0.199	0.029		0.041		0.053	
Ti <sub>2</sub> AsC	0.216				0.171	0.029		0.052			
Ti <sub>2</sub> SC	0.215				0.165	0.027		0.058		0.031	
Ti <sub>2</sub> AlN	0.179				0.153	0.079		0.038			0.069
V <sub>2</sub> AlC	0.205				0.152	0.070		0.037			0.072
Nb <sub>2</sub> AlC	0.150				0.110	0.022					0.060
Cr <sub>2</sub> AlC	0.197				0.153	0.049		0.022			0.074
Ta <sub>2</sub> AlC	0.209				0.154	0.082		0.062			0.067
	M2-C	C-M1				M2		M1	M1-M2		
Ti <sub>3</sub> AlC <sub>2</sub>	0.204	0.219			0.158	0.023		0.069	0.040		0.062
Ti <sub>3</sub> SiC <sub>2</sub>	0.197	0.230			0.175	0.033		0.045	0.037		0.030
Ti <sub>3</sub> GeC <sub>2</sub>	0.192	0.234			0.157	0.037		0.043	0.037		0.029
α-Ta <sub>3</sub> AlC <sub>2</sub>	0.216	0.206			0.145	0.020		0.096	0.040		0.068
α-Ta <sub>4</sub> AlC <sub>3</sub>	C2-Ta2	Ta2-C1		C1-Ta1	0.143	Ta2		Ta1	Ta2-Ta2	Ta2-Ta1	0.068
	0.216	0.218		0.206		0.021		0.095	0.025	0.044	
Ta <sub>5</sub> AlC <sub>4</sub>	Ta3-C2	C2-Ta2	Ta2-C1	C1-Ta1	0.143	Ta3	Ta2	Ta1	Ta3-Ta2	Ta2-Ta1	0.066
	0.219	0.212	0.216	0.209		0.020	0.022	0.096	0.029	0.042	



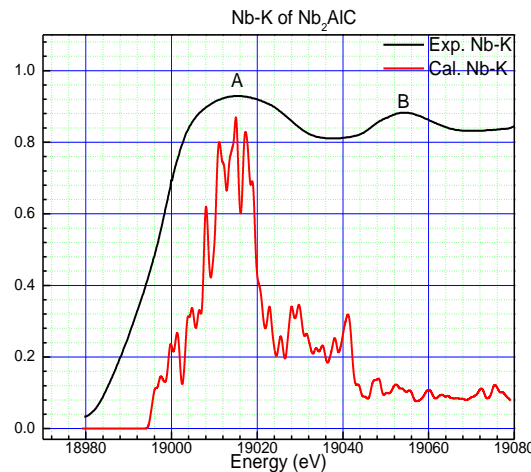
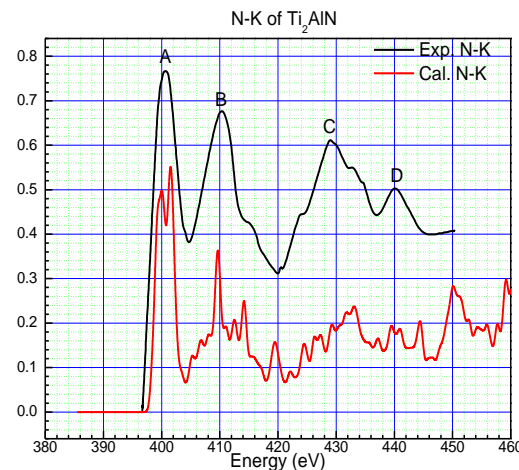
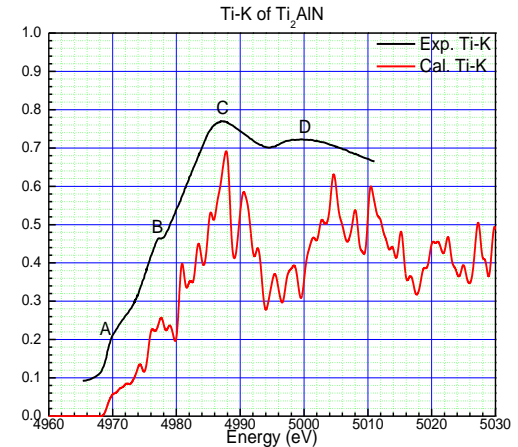
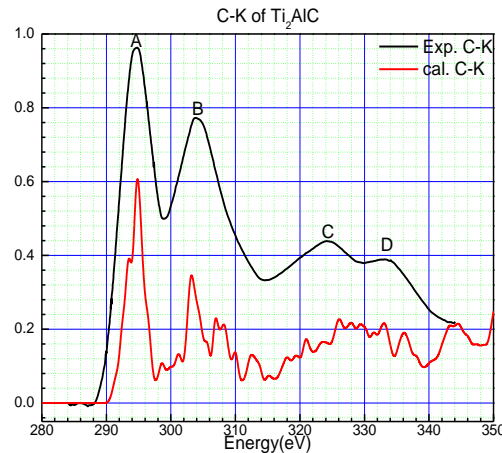
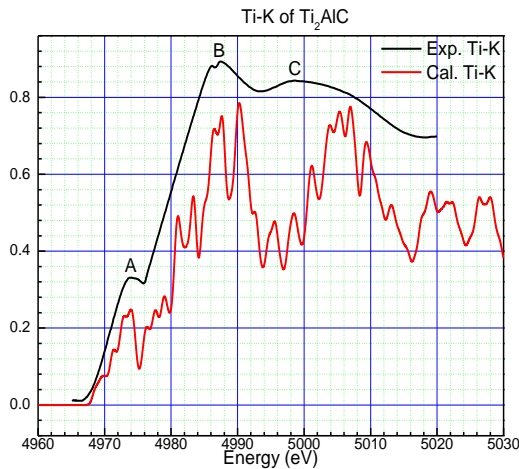
# Optical Conductivities of 20 MAX phases



Thin Solid Films 2009;517:2920

Yu, G., et al., *Photoconductivity and optical conductivity in lightly doped  $Nd_2CuO_{4-\delta}$*  Physica C: Superconductivity, 1992. 203(3-4): p. 419-425

**XANES spectra in MAX phases: Calculated (lower red curves) and measured (upper blue curves) for Ti-K, C-K in  $Ti_2AlC$ , Ti-K, N-K in  $Ti_3AlN$ , and Nb-K edges in  $Nb_2AlC$  crystals. Experimental data from: *G. Hag, M. Jaouen and M. W. Barsoum, Phys. Rev. B 71, 024105, (2005).***



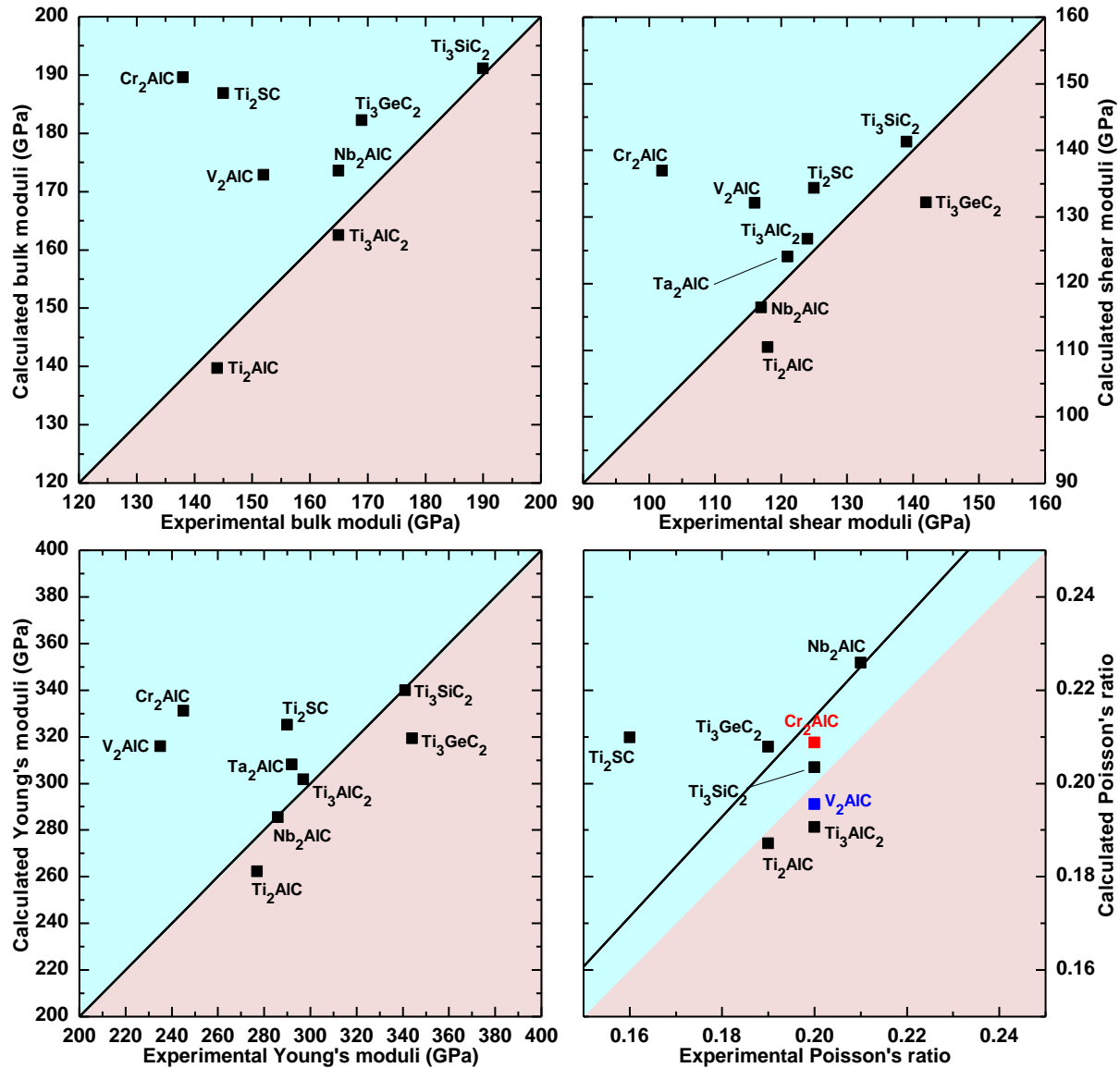
Agreement between calculation and experiment quite good. XANES is useful for characterizing different phases and impurity contaminations. This work is part of the electronic structure results.

# Calculated elastic coefficients and mechanical properties of 20 MAX phases

Crystals	$C_{11}$	$C_{12}$	$C_{13}$	$C_{33}$	$C_{44}$	$C_{66}$	K	G	E	$\eta$	G/K=k
Ti <sub>3</sub> AlC <sub>2</sub>	355.8	81.4	75.3	293.4	120.3	137.2	162.5	126.7	301.7	0.191	0.78
Ti <sub>3</sub> SiC <sub>2</sub>	369.6	96.2	107.6	358.3	155.0	136.7	191.1	141.3	340.0	0.204	0.74
Ti <sub>3</sub> GeC <sub>2</sub>	362.0	97.2	97.7	332.0	137.3	132.4	182.2	132.2	319.3	0.208	0.73
Ti <sub>2</sub> AlC	301.9	68.0	63.0	267.9	105.1	117.0	139.7	110.5	262.3	0.187	0.79
Ti <sub>2</sub> GaC	300.8	79.2	63.8	246.5	92.4	110.8	139.3	101.4	244.9	0.207	0.73
Ti <sub>2</sub> InC	284.4	69.3	55.2	235.5	83.9	107.5	128.6	96.0	230.5	0.201	0.75
Ti <sub>2</sub> SiC	312.9	82.1	110.4	329.2	149.6	115.4	173.0	124.9	302.0	0.209	0.72
Ti <sub>2</sub> GeC	296.6	85.7	96.8	297.1	121.5	105.5	161.0	110.0	268.8	0.222	0.68
Ti <sub>2</sub> SnC	262.6	88.6	73.1	255.2	96.8	87.0	138.8	92.4	226.8	0.228	0.67
<b>Ti<sub>2</sub>PC</b>	<b>256.8</b>	144.8	155.0	<b>339.5</b>	166.3	56.0	191.8	93.1	240.4	<b>0.291</b>	<b>0.49</b>
<b>Ti<sub>2</sub>AsC</b>	<b>212.9</b>	180.4	123.7	<b>289.5</b>	146.3	16.2	150.7	57.2	152.3	<b>0.332</b>	<b>0.38</b>
Ti <sub>2</sub> SC	339.8	101.4	109.7	361.9	159.5	119.2	186.8	134.4	325.2	0.210	0.72
Ti <sub>2</sub> AlN	312.9	73.0	95.5	290.7	126.1	120.0	160.5	117.4	283.1	0.206	0.73
V <sub>2</sub> AlC	334.4	71.5	106.0	320.8	149.8	131.5	172.9	132.1	315.9	0.196	0.76
Nb <sub>2</sub> AlC	316.6	86.3	117.0	288.6	137.6	115.2	173.6	116.4	285.5	0.226	0.67
Cr <sub>2</sub> AlC	366.3	85.8	111.3	356.9	142.9	140.2	189.6	137.0	331.2	0.209	0.72
Ta <sub>2</sub> AlC	344.5	112.2	137.1	327.9	152.3	116.1	198.8	124.1	308.1	0.242	0.62
$\alpha$ -Ta <sub>3</sub> AlC <sub>2</sub>	453.6	130.5	135.6	388.4	175.0	161.5	232.8	161.1	392.8	0.219	0.69
$\alpha$ -Ta <sub>4</sub> AlC <sub>3</sub>	459.2	149.1	148.7	383.1	170.5	155.0	243.0	155.3	384.1	0.237	0.64
Ta <sub>5</sub> AlC <sub>4</sub>	481.5	149.6	158.1	423.6	188.8	165.9	257.2	169.1	416.0	0.231	0.66

Anisotropic ratio  $C_{33}/C_{11}$  correlates with G/K. Note: **Ti<sub>2</sub>PC** and **Ti<sub>2</sub>AsC** are outliers.

# Comparison of our calculated K, G, E, and $\eta$ with the experimental data from *(Barsoum & Radovic, Annu. Rev. Mater. Res. 41, 195, 2011)*.

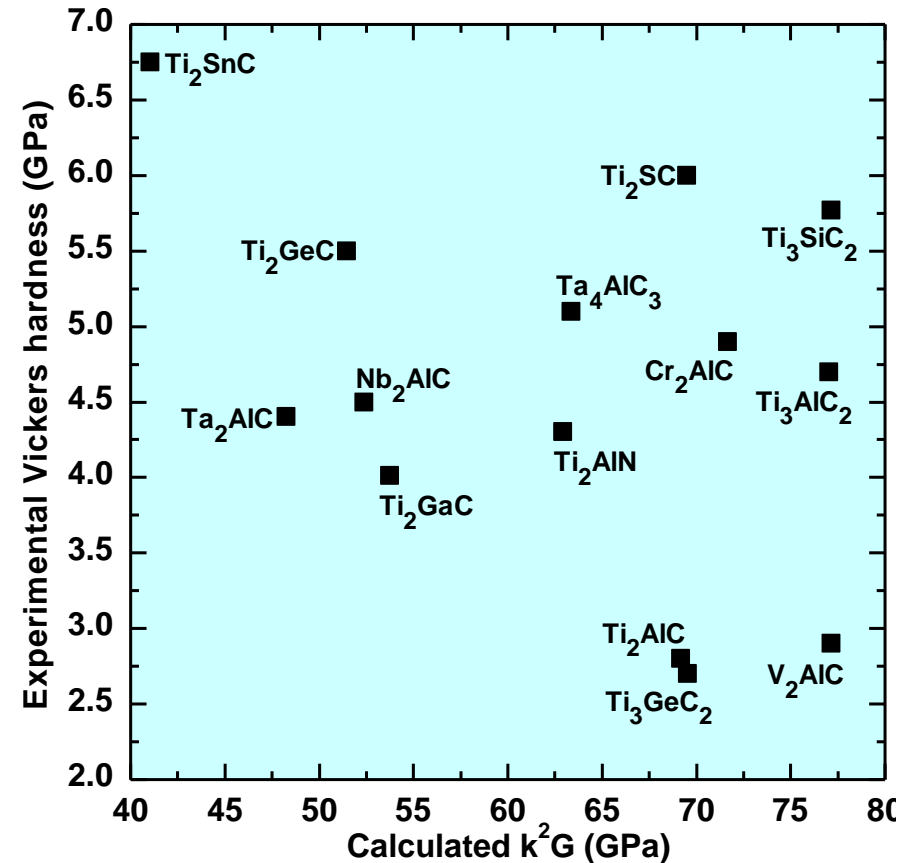


# Implication of our results.

Is it possible to predict experimentally measurable parameters from first-principles calculations? **Yes** and **NO!**

Is there a correlation of Vickers hardness with  $k^2G$ ? Answer: **NO!** (See Figure)

Is it possible to provide theoretical guidance regarding materials development? **Yes.**



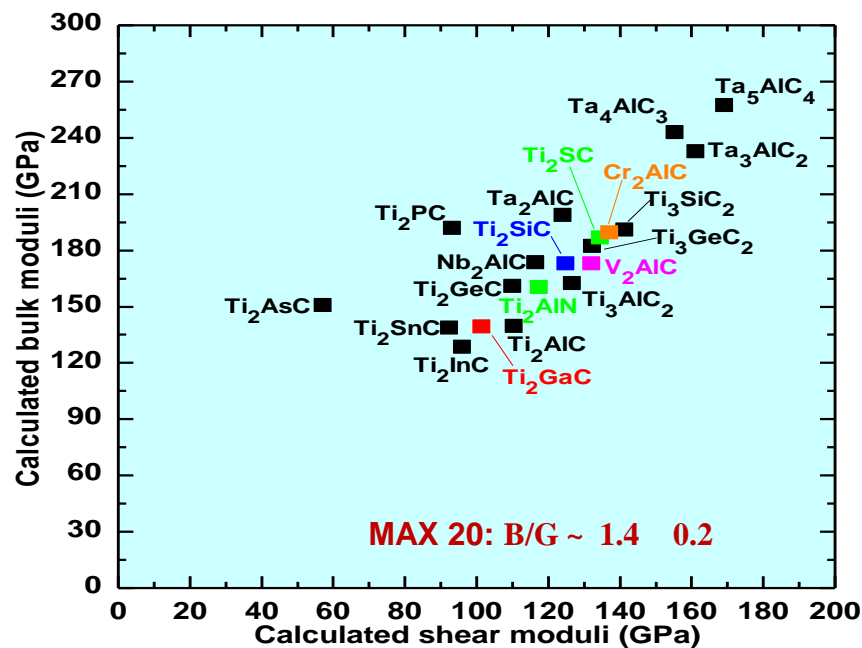
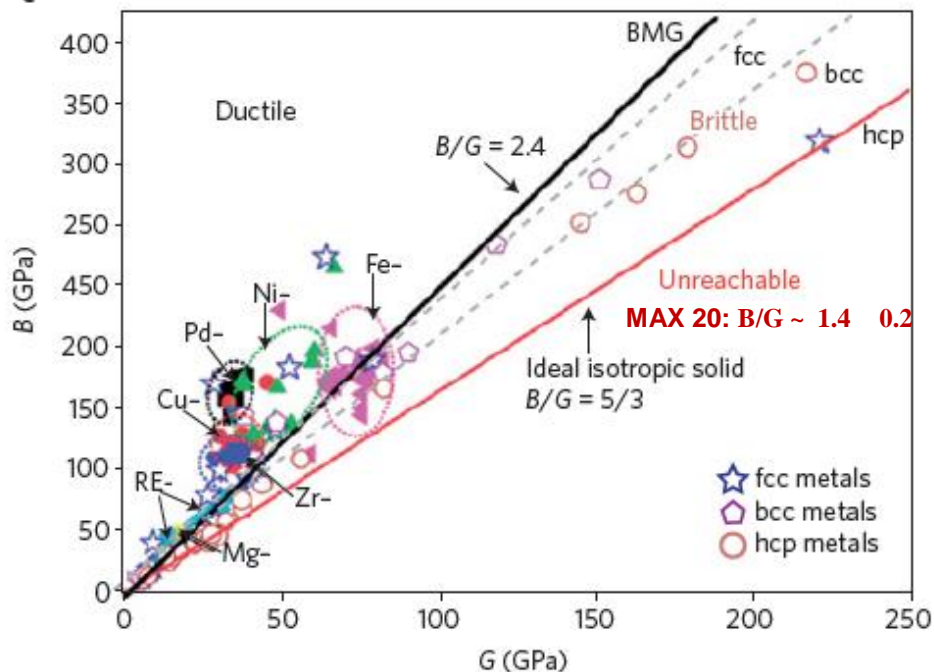
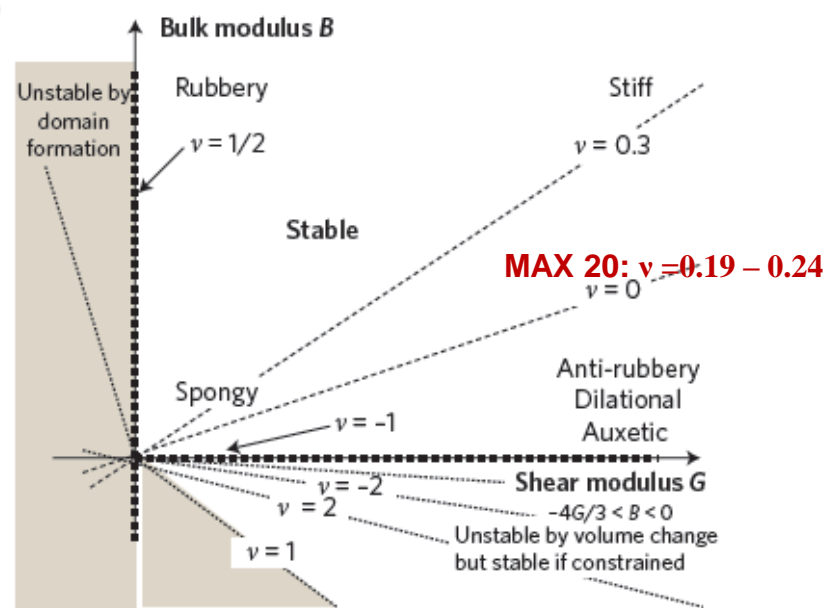
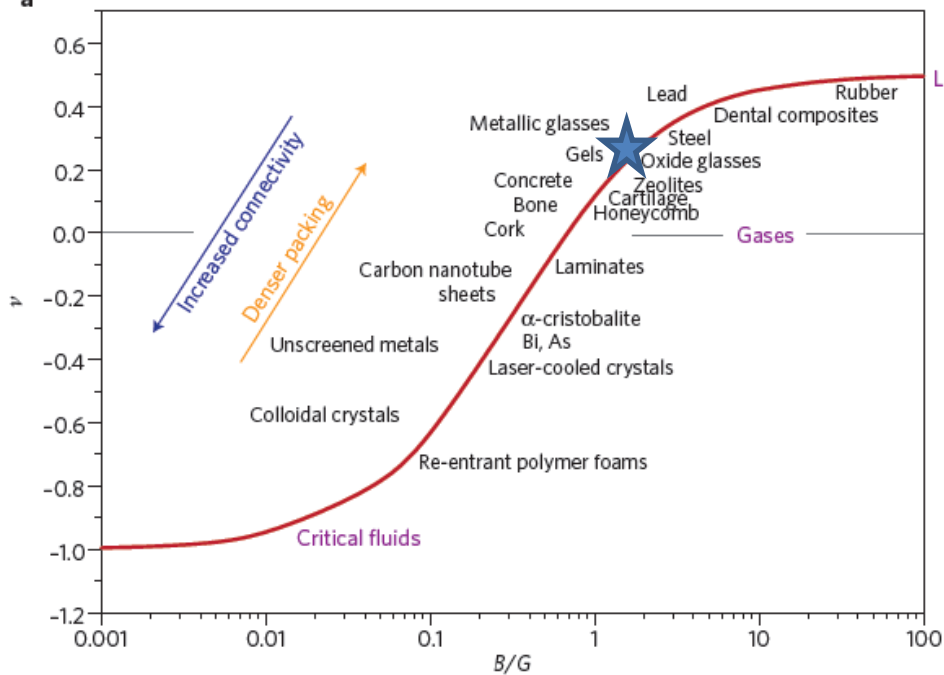
There have been many attempts to predict hardness, failure behavior or other mechanical properties empirically using either experimental or calculated data.

**Results are mixed.**

For **super-hard materials** with simple crystal structures, they are more successful. For **bulk metallic glasses**, more difficult. For **MAX phases**, nearly impossible.

**Reason:** too many factors involved, and complexity of the structure and composition. Still, we will **attempt** through fundamental studies and by understanding atomistic level interactions.

Recent empirical formula by Chan (*Intermetallic, 2011*):  $H_v = 2(k^2G)^{0.585} - 3$  for polycrystalline materials.  $H_v = 0.151 G$  for bulk metallic glasses.



From: G.N. Greaves et al, NM 10, 823, 2011.

# Summary

- The electronic structure, interatomic bonding, interband optical properties, elastic and mechanical properties of 20 MAX phases compounds are calculated by first-principles methods.
- Some of these properties are correlated with experimentally measured data quite well and appear to be able to explain most of their properties based on fundamental electronic structure.
- However, prediction of some properties such as hardness is nearly impossible. MAX phases are a diverse group of layered compounds with considerable variations in their electronic and bonding properties.
- $N(E_f)$  has an approximately linear correlation with electrical conductivity values (except  $\text{Cr}_2\text{AlC}$  and  $\text{Ti}_2\text{InC}$ ). The local feature of total DOS density near  $E_f$  can be used to predict structural stability.
- Effective charge calculation shows that M atoms always lose charge to the X (C or N) atoms whereas A-group atoms mostly gain charge.
- Bond order values are obtained and analyzed for all types of interatomic bonds in all the 20 MAX phases.
- Calculated anisotropic optical conductivities used to predict the anisotropy of electrical conductivities.



## Work in progress and/or contemplated (**wish list?**)

- Predict or correlate hardness and toughness for MAX phases.
- Study magnetic properties of some MAX phases containing the later TM elements ( $\text{Cr}_2\text{AlC}$ ,  $\text{Cr}_2\text{GeC}$ , etc.)
- Extend the list of MAX phases studied (to 60 phases) and build a more complete data base for correlated study. (Materials Genome concept.)
- Explore other properties, thermo-conductivities, electric conductivities, defects, interfaces and corrosion resistance etc.
- Explore the temperature and pressure dependent properties.
- Explore formation of composites in optimizing its properties.
- Explore new unique applications beyond those already recognized.
- Collaborative teaming with others researchers, especially experimentalists.
- Computational simulation for failure behavior => beyond elastic theory and polycrystalline assumption used in the RVH approximation.
- ..... much more.



*Thank you!*